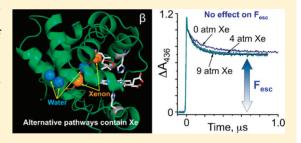


Modulating Distal Cavities in the α and β Subunits of Human HbA Reveals the Primary Ligand Migration Pathway

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ABSTRACT: The free volume in the active site of human HbA plays a crucial role in governing the bimolecular rates of O_2 , CO, and NO binding, the fraction of geminate ligand recombination, and the rate of NO dioxygenation by the oxygenated complex. We have decreased the size of the distal pocket by mutating Leu(B10), Val(E11), and Leu(G8) to Phe and Trp and that of other more internal cavities by filling them with Xe at high gas pressures. Increasing the size of the B10 side chain reduces bimolecular rates of ligand binding nearly 5000-fold and inhibits CO geminate recombination due to both reduction of the capture volume in the distal pocket and direct steric hindrance of Fe—ligand bond formation. Phe and Trp(E11) mutations also cause a decrease in



distal pocket volume but, at the same time, increase access to the Fe atom because of the loss of the $\gamma 2$ CH₃ group of the native Val(E11) side chain. The net result of these E11 substitutions is a dramatic increase in the rate of geminate recombination because dissociated CO is sequestered close to the Fe atom and can rapidly rebind without steric resistance. However, the bimolecular rate constants for binding of ligand to the Phe and Trp(E11) mutants are decreased 5–30-fold, because of a smaller capture volume. Geminate and bimolecular kinetic parameters for Phe and Trp(G8) mutants are similar to those for the native HbA subunits because the aromatic rings at this position cause little change in distal pocket volume and because ligands do not move past this position into the globin interior of wild-type HbA subunits. The latter conclusion is verified by the observation that Xe binding to the α and β Hb subunits has little effect on either geminate or bimolecular ligand rebinding. All of these experimental results argue strongly against alternative ligand migration pathways that involve movements through the protein interior in HbA. Instead, ligands appear to enter through the His(E7) gate and are captured directly in the distal cavity.

The existence of internal cavities in proteins was first The existence of internal cavince in production of discovered in sperm whale myoglobin. In 1965, Kendrew's group and then, in 1985, Petsko and co-workers reported crystal structures of Mb containing bound Xe. These Xe sites are found in the proximal part of the heme pocket just below pyrrole ring B (called Xe1), behind the heme next to the vinyl group of pyrrole B (Xe2), in the back of the distal pocket (Xe4), and between the EF corner and the H helix (Xe3). Later, it was demonstrated that these Xe cavities are involved in the kinetics of internal ligand movement after photolysis of Mb-ligand complexes. 3-8 Molecular modeling and dynamics simulations went further and predicted that these apolar cavities may function as transient stations for ligands migrating from the solvent to the active site in Mb. 9-14 However, most experimental data, including exhaustive mutagenesis mapping and time-resolved X-ray crystallography, do not support these theoretical conclusions and suggest, instead, that ligands enter and escape Mb by the His(E7) gate pathway. 3,4,15-21 Xe cavities are also present in the interior of HbI from the blood clam Scapharca inequivalvis (HbI), can trap photodissociated CO, and have been proposed to be part of the ligand migration pathway in this mollusck globin. However, blocking these docking sites with Xe, organic halides, and space-filling Phe and Trp mutations causes no appreciable change in the fraction of escape of ligand from the protein. Instead, ligands appear to

enter and escape the clam Hb through the distal histidine gate. 22,23

However, the His(E7) channel is not the universal pathway for the movement of ligand into all globins. In some bacterial truncated hemoglobins and the neuronal mini-globin from the *Nemertean* sea worm *Cerebratulus lacteus*, ligands enter the protein through an apolar tunnel between the C-terminal ends of the E and H helices.²⁴ In *C. lacteus* Hb (CerHb), either blocking the apolar tunnel with Trp mutations or filling the tunnel with Xe markedly decreases both the fraction of escape of ligand to solvent and the bimolecular rate constant for entry of ligand into the active site.^{25,26}

There is much less experimental information regarding the functional role of internal cavities in human HbA subunits. Adachi et al. 27 demonstrated that after CO photodissociation, an electron density feature corresponding to the ligand was observed in the primary docking site, located 3.5 Å from the Fe atom and above the pyrrole C ring in α subunits of human HbA. In β subunits, a similar primary photodissociated site is occupied, and a second small ligand electron density peak is found in the back of the distal pocket, ~8.5 Å from Fe, in a

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cavity circumscribed by the B10, E11, E12, and G8 amino acid side chains.

Sottini et al. 28,29 measured CO rebinding kinetics in R-state HbCO that was encapsulated in silicon hydrogels in the presence of 80% glycerol and observed two geminate phases with characteristic rebinding times of $\sim 10^{-8}$ and 10^{-6} s. These phases were assigned to recombination from two different internal sites. On the basis of molecular modeling studies, the authors proposed that the slower geminate processes correspond to rebinding from more internal cavities, which was observed in Mb. 3 To locate these internal, apolar cavities, Savino et al. 30 crystallized deoxyHbA under a high pressure of Xe gas. Several Xe atoms were modeled into the electron density with some positions corresponding spatially to the Xe4 and Xe2 cavities in Mb. Savino et al. 30 proposed that these Xe binding sites map multiple ligand migration pathways from the active site to the solvent in both subunits of human HbA.

In our previous work, we attempted to shut the E7 gate in HbA by replacing the distal His with a larger Trp residue. 31,32 The bimolecular rates of binding of CO and O₂ to the blocked and closed Trp(E7) conformers in HbA decreased ≥10-fold in comparison to those for wild-type (WT) HbA, suggesting that ≥90% of ligands enter the active site via the His(E7) gate. However, we did not explore the role of distal pocket volume and internal cavities in regulating bimolecular ligand binding. In an attempt to reconcile the discrepancy between our mutagenesis work at the E7 position and the Xe binding and theoretical work of Sottini et al. 28,29 and Savino et al., 30 respectively, we have systematically measured the effects of high Xe pressures and Phe and Trp substitutions at the internal B10, E11, and G8 positions on binding of ligand to isolated α and β subunits and tetrameric HbA. The results indicate that in native HbA ligands are sequestered near the Fe atom after photodissociation, do not migrate into internal cavities, and escape and enter the distal pocket by the E7 gate, a pathway that appears to be universal for vertebrate myoglobins and red cell hemoglobins.

MATERIALS AND METHODS

Recombinant Hb (rHb) Mutagenesis, Expression, and Purification. The original plasmid system for expressing rHb with V1M mutations in each subunit and the bacterial strain for production with added external hemin were developed at Somatogen, Inc.³³ Mutagenesis was performed using Stratagene's polymerase chain reaction-based "QuikChange Site-Directed Mutagenesis" kit (Stratagene, La Jolla, CA). Cell growth and protein expression protocols were described previously by Birukou et al.³¹ Protein was isolated and purified according to the method of Looker et al.³³ Individual Hb subunits were obtained following the procedure described by Birukou et al.³¹ The purity of isolated Hb subunits was assessed by cellulose acetate electrophoresis (Helena Laboratories, Beaumont, TX).

Laser Photolysis of Hb Subunits in O_2/CO Mixtures. Association rate constants for binding of O_2 and CO to the mutant Hbs (k'_{O_2} and k'_{CO} , respectively) were measured using laser-flash photolysis of the stable HbCO complexes in O_2/CO mixtures ranging from $\sim 100\%$ O_2 to 100% CO. Photolysis was initiated by a 0.5 μ s excitation pulse generated by a phase-R 2100B dye laser system using Rhodamine 575. The initial rapid, bimolecular phases were used to determine k'_{O_2} and k'_{CO} as described by Birukou et al.³¹ and Salter et al.²⁵ In selected cases,

the k'_{O_2} values were determined by photolysis of the HbO₂ complex.

NO Binding and NO Dioxygenation. Parameters for bimolecular NO binding to isolated mutant α and β subunits were obtained using a flow-flash method, in which an anaerobic solution of HbCO with no excess CO was mixed with high concentrations of NO and then immediately photolyzed to allow complete binding of NO to the transiently formed deoxyHb as described by Birukou et al. 31 The rate constants for NO dioxygenation by HbO2 complexes were measured by mixing 2 μ M solutions of oxygenated forms of α and β subunits with 10 µM solutions of NO in a modified Gibson-Durrum stopped-flow spectrophotometer D-110 (originally from Dionex Instrument, Palo Alto, CA). Absorbance changes were measured at 402 nm. Experiments were conducted in 100 mM sodium phosphate (pH 7) at 20 °C. A detailed description of the preparation of gas-saturated solutions of NO, O₂, and CO can be found in ref 31, and the protocols for NO dioxygenation have been described previously. 34,38

Geminate CO Recombination. Time courses for geminate CO recombination within isolated native and mutant HbCO subunits were collected after excitation with a 7 ns pulse from a frequency-doubled YM600 Nd:YAG laser system (Lumonics, Inc., Billerica, MA). The sample was illuminated by a pulsed Xe arc lamp (model 03-102, Applied Photophysics, Inc., Leatherhead, U.K.) to increase the "signal-to-noise" ratios of the absorbance traces. Transmittance was measured at 436 nm using a photomultiplier tube with a 0.9 ns response time (R-1913, Hamamatsu, Japan); 32 traces were averaged for each measurement. All measurements were performed at 20 °C in 100 mM sodium phosphate (pH 7). Several grains of sodium dithionite were added to each cuvette for the HbCO complexes to remove residual O_2 and prevent Hb oxidation.

Xe Binding Experiments. The method for adding high pressures of Xe gas was adopted from refs 4 and 26. Bimolecular O2 and CO binding and geminate CO recombination to HbA tetramers and isolated subunits were measured in the presence of 0, 4, and 9 atm of Xe in a stainless steel pressure cell constructed originally by Q. H. Gibson. ⁴ A small amount of Hb sample (0.5-1 mL) equilibrated with 1 atm of CO (1000 μ M CO) or 1 atm of air (~250 μ M O₂) was introduced into the pressure cell, which was sealed with a rubber septum and purged with CO or air, respectively, in advance of pressurization. For CO samples, several grains of dithionite were added to the cell before the introduction of the sample. Initially, time courses for O2 and CO binding were collected in the absence of Xe, either in air or in pure CO, both at 1 atm pressure. Then, Xe was introduced via a three-way valve attached to the cell, allowing no gas from the original sample to escape. The solution in the cuvette was equilibrated with the Xe/CO or Xe/air mixture by extensive shaking of the cuvette for several minutes. The final total pressures of the gas mixtures in the cell were 1, 5, and 10 atm. Bimolecular and geminate rebinding measurements were conducted as described above, within 10-20 min of equilibration with Xe.

RESULTS

Nanosecond CO Recombination to Native α and β Subunits and HbA Tetramers. Geminate rebinding after laser photolysis provides indirect evidence of the extent of ligand migration that occurs inside globins immediately after

photolysis of the Fe-ligand bond. Depending on the reactivity of the ligand and Fe, the length of the laser pulse, and the experimental conditions, rebinding from the initial docking site of the ligand and from more distant locations within the protein can be observed. Compared to Mb where the rate of internal O_2 rebinding is moderate, ~10 μ s⁻¹, O₂ geminate recombination in HbA is very fast (>100 μ s⁻¹) because of the high reactivity of the heme Fe³⁶ and reflects primarily ligand rebinding from the vicinity of the Fe atom.³⁷ More information about ligand migration in HbA subunits can be obtained by measuring geminate CO recombination, because the rate of formation of the Fe-CO bond is ~10-fold smaller than that for the Fe-O₂ bond, allowing migration of CO to more distant sites. In the case of Mb, the low reactivity of the iron atom is due to constraints on in-plane movement of the proximal histidine. These restrictions decrease the rate of internal CO recombination in Mb to $\leq 0.1 \ \mu s^{-1}$, ³⁸ and as a result, there is little or no geminate recombination at room temperature $(\leq 5\%)$. Thus, the effects of mutagenesis on ligand migration in Mb can be examined only under physiological conditions using O_2 .

Figure 1 shows the time course for geminate CO recombination within a native human HbA tetramer. The

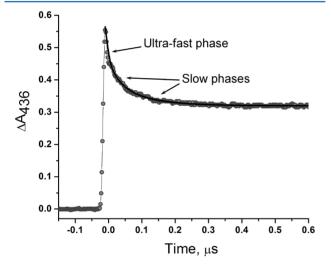


Figure 1. Ultrafast and slow phases for CO geminate recombination within tetrameric native HbA. A solution of $\sim\!100~\mu\mathrm{M}$ HbA(CO) in 1000 $\mu\mathrm{M}$ CO was photolyzed with a 7 ns Nd:YAG laser excitation pulse, and absorbance changes were monitored at 436 nm. The ultrafast phase represents ligand rebinding from the initial photolyzed complex, which occurs on time scales of less than $\sim\!5-10$ ns. The slower geminate phase represents rebinding within the distal pocket. Measurements were conducted in 100 mM sodium phosphate (pH 7) at 20 °C.

trace is clearly biphasic, with one ultrafast phase followed by a slower geminate rebinding process, which occurs with a half-time of 50–100 ns. The ultrafast rebinding phase cannot readily be resolved using our 7 ns YAG laser because it decays simultaneously or even faster than the laser pulse at a rate of >100 μ s⁻¹. A representation of the events occurring after the laser flash is provided in Scheme 1 and based on previous interpretations of geminate rebinding in HbA subunits and Mb. 4,15,37,39,40 The ultrafast phase corresponds to ligand recombination from an initial B or first geminate state, presumably representing ligand transiently located just above

Scheme 1. General Four-State Geminate Rebinding

HbX
$$\xrightarrow{k_1} B \xrightarrow{k_2} C_1, C_2, \dots \xrightarrow{k_4} Hb + X$$

the Fe atom. The existence of this initial "docking" site was previously identified in Mb^{40} and suggested for hemoglobin subunits.³⁷ The slower geminate phase represents recombination of ligands in rapid equilibrium between state B and locations C_1 , C_2 , etc., which are more distant from the Fe but still within or adjacent to the distal pocket.

When rebinding and escape from the B state (k_1 and k_2 , respectively) are significantly faster than geminate rebinding from the C states, the mechanism can be simplified to a two-step scheme with a single geminate intermediate, C, which represents rebinding by ligands that are in or near the distal pocket (Scheme 2). In terms of Scheme 1, the rate of formation of the ligand—Fe bond can be expressed as $k_{\text{bond}} = k_3 k_1/(k_1 + k_2)$.

Scheme 2. Two-Step Geminate Rebinding for Nanosecond Geminate Recombination

$$HbX \xrightarrow{k_{bond}} C \xrightarrow{k_{escape}} Hb + X$$

Scheme 2 allows calculation of the rates of ligand bond formation $(k_{\rm bond})$, escape from the protein $(k_{\rm escape})$, and entry $(k'_{\rm entry})$ using (a) the fitted fraction of geminate recombination $[F_{\rm gem} = k_{\rm bond}/(k_{\rm bond} + k_{\rm escape})]$, (b) the observed rate of recombination $(k_{\rm gem} = k_{\rm bond} + k_{\rm escape})$, and (c) the overall bimolecular association rate constant for CO rebinding from solvent $[k'_{\rm CO} = (k'_{\rm entry}k_{\rm bond})/(k_{\rm bond} + k_{\rm escape})]$, which is measured on much longer time scales. ¹⁵

$$k_{\text{bond}} = k_{\text{gem}} F_{\text{gem}} \tag{1}$$

$$k'_{\text{entry}} = \frac{k'_{\text{CO}}}{F_{\text{gem}}}$$
 (2)

$$k_{\text{escape}} = k_{\text{gem}} (1 - F_{\text{gem}}) \tag{3}$$

Time courses for rebinding of geminate CO to isolated α and β subunits of HbA are shown in Figure 2A, and as with tetrameric HbA, a small ultrafast phase is seen, with roughly the same amplitude for all three proteins. In contrast, there are significant differences between the subunits on longer time scales. Fitted parameters for the α and β slow geminate phases are listed in Table 1 and agree well with previous measurements using a longer excitation pulse.³⁷ Nanosecond geminate rebinding is faster in α (20 μ s⁻¹) than in β subunits (7 μ s⁻¹), whereas the fraction of recombination is larger in β (0.26) than in α subunits (0.16) (Table 1).

When the subunits are present in tetramers, all the geminate parameters increase. $F_{\rm gem}$ becomes ~0.30 for both subunits due to 2–3-fold increases in $k_{\rm bond}$ (Table 1). In contrast, there is very little difference in $k_{\rm escape}$ values for tetramers versus isolated subunits, and Birukou et al.³¹ showed that there is also very little difference between the $k'_{\rm entry}$ values for isolated and tetrameric subunits. Thus, the increase in $k_{\rm bond}$ for the subunits in tetramers suggests that the Fe atom is more reactive, perhaps due to fixing the proximal heme—His(F8) complex in a

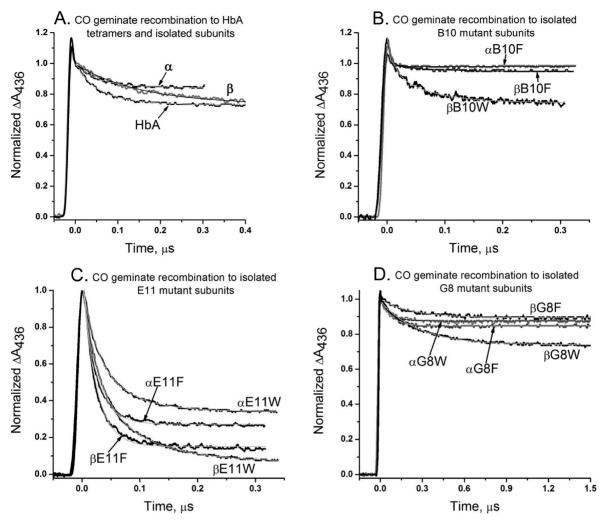


Figure 2. CO geminate rebinding to WT and mutant Hbs. The protein concentrations were kept between 50 and 100 μ M, and the CO concentration was 1000 μ M. Rates and amplitudes of slow geminate rebinding to subunits within tetramers were obtained by fitting the slow phases to a two-exponential expression $[A(t) = A_{\text{escape}} + A_{\text{gem},\alpha} \exp(-k_{\text{gem},\alpha}t) + A_{\text{gem},\beta} \exp(-k_{\text{gem},\beta}t)$, whereas the parameters for slow geminate rebinding in isolated subunits were fitted to a single-exponential expression $[A(t) = A_{\text{escape}} + A_{\text{gem}} \exp(-k_{\text{gem},\beta}t)$, whereas the parameters for slow geminate rebinding in isolated subunits were fitted to a single-exponential expression $[A(t) = A_{\text{escape}} + A_{\text{gem}} \exp(-k_{\text{gem}}t)$. The smooth gray lines represent the best-fit exponential expression. The time courses were normalized by dividing each observed absorbance change point by the total change measured at 436 nm (excluding the ultrafast phase). F_{gem} was obtained by dividing A_{gem} by the total ΔA_{436} (or $0.5\Delta A_{436}$ for α or β subunits within a tetramer, because only 50% of the total absorbance change of the tetramer corresponds to a particular subunit). k_{gem} is equal to the observed rate of the first-order internal rebinding phase. (A) Geminate CO recombination within HbA tetramers and isolated native α and β subunits. (B) Geminate CO recombination within isolated B10 mutant subunits (B10 corresponds to position 29 in α subunits and 28 in β subunits). (C) Geminate CO recombination within isolated G8 mutant subunits (E11 corresponds to position 62 in α subunits and 106 in β subunits). Conditions were the same as those described for Figure 1.

staggered orientation because of interactions with partner subunits. In contrast, the F helices in isolated subunits are probably more flexible. This interpretation is supported by the observation that the α subunits show the largest increase in $k_{\rm bond}$, presumably because these subunits are almost completely monomeric at the protein concentrations used (50–100 μ M), whereas under these conditions, isolated β subunits are primarily homotetrameric.

Geminate and Bimolecular Ligand Binding to B10 Mutants. Representative time courses for geminate CO recombination in α Phe(B10), β Phe(B10), and β Trp(B10) isolated subunits are shown in Figure 2B. The ultrafast phase, reflecting ligand rebinding from the initial photodissociated B state, was observed in the time courses for each mutant subunit, but then little or no slow nanosecond recombination is

observed for these variants, with the exception of $\beta Trp(B10)$. The $\alpha Trp(B10)$ mutant is too unstable to be prepared as an isolated chain. Consequently, the hybrid $\alpha (Trp(B10))\beta (WT)$ tetramer was used to measure parameters for the $\alpha Trp(B10)$ subunit. The fitted parameters for the slow nanosecond geminate phases of the B10 mutants are listed in Table 2, and F_{gem} represents the fraction of the slower geminate phase.

In α subunits, both the Phe(B10) and Trp(B10) mutations cause complete inhibition of nanosecond geminate rebinding. In β Phe(B10) subunits, the fraction of nanosecond geminate recombination is reduced to 4%. These results are consistent with the effects of Phe and Trp(B10) replacements on geminate O₂ rebinding to Mb, where $F_{\rm gem}$ is reduced from 0.5 for wild-type Mb to 0.3 and 0.05 for the Phe and Trp(B10) mutants, respectively.³ Pronounced reduction of nanosecond

Table 1. Geminate Parameters of CO Recombination to HbA Tetramers and Isolated Subunits

subunit	parameter ^a	isolated subunit	tetramer
αWT	$k_{\rm gem}~(\mu {\rm s}^{-1})$	20 ± 1	34 ± 3
	$F_{ m gem}$	0.16 ± 0.01	0.28 ± 0.06
	$k_{\mathrm{bond}} \ (\mu \mathrm{s}^{-1})$	$3.1 \pm 0.4 (2.5 \pm 1.0)^b$	9.4 ± 1.3
	$k_{\mathrm{escape}} \ (\mu \mathrm{s}^{-1})$	$17 \pm 2 (13 \pm 4)^b$	24 ± 4
β WT	$k_{\rm gem}~(\mu {\rm s}^{-1})$	7.1 ± 0.6	10 ± 2
	$F_{ m gem}$	0.26 ± 0.02	0.33 ± 0.03
	$k_{\mathrm{bond}} \ (\mu \mathrm{s}^{-1})$	$1.9 \pm 0.3 (1.7 \pm 0.4)^b$	3.5 ± 0.8
	$k_{\rm escape}~(\mu { m s}^{-1})$	$5.3 \pm 0.4 \ (4.8 \pm 1.0)^b$	7.0 ± 1

^αThe geminate parameters were computed using Scheme 2 and eqs 1–3. To compare $F_{\rm gem}$ values of isolated subunits to those of subunits within HbA tetramers, $F_{\rm gem}$ values for the α and β phases in the tetramer data were divided by 0.5 to normalize for heme concentration (i.e., in HbA one-half of heme groups are α and the other half are β). ^bThe values in parentheses were taken from ref 37 and were computed by reducing Scheme 1 to Scheme 2 as described in the text [i.e., $k_{\rm bond} = (k_3k_1)/(k_1 + k_2)$, and $k_{\rm escape} = k_4$].

geminate CO rebinding was also reported for tetrameric Hb mutants containing Tyr(B10) and Gln(E7) in both subunits 41 and hybrid Hb tetramers containing $\alpha \mathrm{Phe}(\mathrm{B10})$ or $\alpha \mathrm{Trp}(\mathrm{B10})$ and allosteric mutations. 42 These results demonstrate that bulky aromatic side chains at the B10 position severely hinder access of photodissociated ligands to the heme Fe atom in all three globins, decreasing k_{bond} markedly.

Surprisingly, the geminate recombination parameters for $\beta \text{Trp}(\text{B10})$ are very similar to those for WT β subunits (Table 2). The more flexible β distal pocket appears to be able to accommodate both the large Trp(B10) side chain and the bound ligand, and rebinding from the C geminate state is not inhibited by the indole side chain, which appears to have moved away from the active site and is unable to swing back over the Fe atom until the ligand has left the pocket. The biphasic character of bimolecular CO rebinding to this mutant suggests that the TrpB10 side chain in β subunits can occupy at least two distinct conformations, one with the indole ring greatly hindering ligand binding, which is observed for $\alpha \text{Trp}(\text{B10})$, and one in which the B10 side chain is pointing away from the Fe atom (Table 2).

Space-filling mutations at the B10 position also dramatically affect bimolecular ligand binding, particularly for α subunits. The CO association rate constant, k'_{CO} , decreases 160- and 1600-fold for the α Phe(B10) and α Trp(B10) mutants, respectively, compared to that for the WT α subunit. Similar trends are observed for O₂ and NO binding (Table 2, second and third rows, respectively). The large differences between the absolute values of k'_{NO} , k'_{O_2} , and k'_{CO} are due to intrinsic differences in the ligand reactivities with Fe (NO > $O_2 \gg CO$), with the formation of the Fe-ligand bond being rate-limiting for CO binding and entry of ligand into the distal pocket being rate-limiting for NO binding. 34,44 O₂ association is regulated by a combination of these processes. The dramatic inhibition of binding of CO to the α Phe(B10) and α Trp(B10) mutants is caused by severe hindrance of access to the Fe atom by the large phenyl and indole rings. The smaller 6- and 30-fold decreases in k'_{NO} for the α Phe(B10) and α Trp(B10) mutants, respectively, reflect the decrease in size of the distal volume for the initial noncovalent capture of the ligand molecule. The decreases in k'_{O_2} are intermediate between those for CO and

Table 2. Rate Constants for Bimolecular CO, NO, and O₂ Binding and NO Dioxygenation and Parameters for the Slow CO Geminate Recombination to Isolated α and β B10 Mutant Subunits^a

s ⁻¹)	(0)			(6	(44%)	3 (42%)		
$k'_{\rm O_2}~(\mu{\rm M}^{-1}~{\rm s}^{-1})$	$40 \pm 2 (35 \pm 10)$	0.87 (0.64)	0.033 (0.10)	$63 \pm 12 \ (67 \pm 9)$	26 (56%), 5.8 (44%) (8.2)	23 (58%), 0.013 (42%) (0.033)		
13	80^{b}	8.0 (3.0)	1.6^e	80^{b}	$\sim \! 100^f (50)$	50 (72%), 3.7 (28%)		
$K_{\rm entry}~({\rm M}^{-1})~~k'_{\rm NO}~(\mu{\rm M}^{-1}~{\rm s}^{-1})$	31	4.5	0.81	89	57	22 (56%), 0.34 (44%)		
$K_{\rm entry}~({ m M}^{-1})$	2.0 ± 0.2	0.27^{d}	0.048^{d}	8.4 ± 2.0	2.1, 0.35	2.0, 0.0011		
$k'_{ m entry} \; (\mu { m M}^{-1} \; { m s}^{-1})$	33 ± 4	pu	pu	42 ± 13	37, 6.3	23, 0.013		
$k'_{\rm CO}~(\mu{ m M}^{-1}~{ m s}^{-1})$	$5.2 \pm 0.5 \ (2.9 \pm 1.2)$	0.032 (0.020)	0.0033^c	$11 \pm 3.3 \ (6.8 \pm 1.5)$	1.6 (48%), 0.27 (52%) (0.32)	5.1 (62%), 0.0029 (38%) (0.0016)		
$k_{\rm escape} \ (\mu { m s}^{-1})$	17 ± 2	pu	pu	5.3 ± 0.4	18	12		
) $k_{\text{bond}} (\mu s^{-1}) k$	3.1 ± 0.4	pu	pu	1.9 ± 0.3	0.81	3.4		
$k_{\rm gem}~(\mu {\rm s}^{-1})$	20 ± 1	pu	pu	7.1 ± 0.6	19	15		
					0.04	0.22		
subunit	lphaWT	lphaPhe $(B10)$	$a{ m Trp}({ m B}10)^c$	$ ho_{ m MT}$	hoPhe(B10)	$ ho ext{Trp}(ext{B10})$		
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entry and Kentry were calculated or estimated using eqs 1-3. k_{gem} , k_{bond} , and k_{escape} for $lpha \mathrm{Phe}(\mathrm{B10})$ and $lpha \mathrm{Trp}(\mathrm{B10})$ were in parentheses for k'_{CO} and k'_{O_2} are for subunits within tetramers and were taken from ref 43. K_{NOD} values in parentheses for α - and β Phe(B10) variants in tetramers were taken from ref 34. Conditions were the same as those described for Figure 1. $^bK_{NOD}$ for α - and β WT subunits within type a parameter was used for the rate of ligand escape. "The amplitude of the slow phase in both NO dioxygenation and bimolecular NO binding was less than 0.5 of the total absorbance change. The fast ^aThe naturally occurring amino acid at the B10 position is Leu in both subunits. In cases where two bimolecular binding phases were observed for isolated subunits, the relative percentages of each phase HbA tetramers. ^cGeminate and bimolecular binding parameters are for α Trp(B10) subunits within hybrid mutant/WT tetramers. ^d $k_{\rm NO}$ was used as the rate of ligand entry for α Trp(B10), and the wildspaneters for the hybrid HbA tetramers agreed well and were assigned to the wild-type β subunits. The rate of reaction was too fast to be reliably measured. are given in parentheses. Parameter values were rounded to two significant digits. k'not determined (nd) because F_{gem}

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NO, supporting the idea that both ligand entry and ligand—Fe bond formation limit O₂ association.

Bimolecular binding to the $\beta(B10)$ mutants is more complex because of the appearance of an initial fast phase (~50% of the amplitude) with a rate similar to that of WT β subunits and an ultraslow phase similar to that seen for the $\alpha(B10)$ mutants (Table 2). The faster phase almost certainly reflects an alternative, less hindered conformation of the large phenyl and indole B10 side chains. The presence of conformational heterogeneity in the β Phe(B10) and Trp(B10) mutants of Hb was previously noticed in nuclear magnetic resonance studies by Wiltrout et al. 45 The crystal structures of HbA alkyl isocyanide complexes show that the Leu(B10) side chain in β subunits is flexible and can move away from large bound ligand molecules toward Leu(B13), whereas in α subunits, the Leu(B10) side chain is held in place by the larger Met(B13) side chain.³⁴ This flexibility in the β subunits also appears to exist in Phe(B10) and Trp(B10) mutants, as judged by the presence of two bimolecular phases and the lack of a large effect of the aromatic side chains on the fast bimolecular phases for the binding of all three ligands (Table 2).

The rate constants for slow bimolecular ligand binding to $\beta \mathrm{Trp}(\mathrm{B10})$ are identical to those for $\alpha \mathrm{Trp}(\mathrm{B10})$ subunits, suggesting that ~50% of the $\beta \mathrm{Trp}(\mathrm{B10})$ indole side chains are located directly above the heme Fe atom. When rate parameters for ligand binding to the $\beta \mathrm{Phe}(\mathrm{B10})$ and $\mathrm{Trp}(\mathrm{B10})$ mutants were estimated for subunits within mutant/WT hybrid tetramers, only the ultraslow phases could be resolved from the WT subunit binding phases. As shown in Table 2, the k'_{CO} and $k'_{\mathrm{O_2}}$ values for the mutant subunits in tetramers are very similar to those determined for the slow bimolecular phases observed with isolated subunits.

As is the case for simple NO binding to deoxyHb, entry of bimolecular ligand into the distal pocket is also the rate-limiting step for NO dioxygenation by HbO₂ complexes, which is measured by rapidly mixing solutions of α O₂ or β O₂ with varying NO concentrations.³⁴ Thus, k'_{NOD} as well as k'_{NO} can be used to verify the values of k'_{entry} computed from overall and geminate CO binding parameters. As expected, NO dioxygenation is significantly inhibited in α Phe(B10), α Trp(B10), and β Trp(B10) subunits (Table 2), whereas k'_{NOD} and k'_{NO} are little affected or increase slightly for the β Phe(B10) mutant, as observed previously.³⁴

Geminate and Bimolecular Ligand Binding to E11 Mutants. Time courses for geminate CO recombination to Phe and Trp(E11) mutants of the α and β subunits are shown in Figure 2C. There is no distinct ultrafast geminate phase for these variants. Fitted geminate parameters and calculated rates of bond formation and escape are listed in Table 3. Unlike the position B10 substitutions, the Phe and Trp(E11) mutations markedly increase $F_{\text{gem,CO}}$ in HbA, from 0.16 to 0.70 in α mutants and from 0.26 to 0.90 in β mutants. Analogous effects were reported for Phe and Trp(E11) mutants of Mb, where $F_{\text{gem,O2}}$ increased from 0.50 in WT protein to ≥0.90 in the E11 aromatic amino acid variants.

The increases in $F_{\rm gem}$ for the Phe(E11) mutants are due primarily to 10- and 20-fold increases in $k_{\rm bond}$ for the α and β subunits, respectively. The analysis of the Hb Trp(E11) mutants is complicated by the presence of two slow nanosecond geminate phases. However, regardless of the interpretation of these two phases, it is clear that the increase in the rate of geminate recombination is due to markedly

Table 3. Rate Constants for Bimolecular CO, O_2 , and NO Binding, NO Dioxygenation, and CO Geminate Recombination for Isolated α and β E11 Mutant Subunits^{α}

	$k'_{{\rm O}_2}~(\mu{ m M}^{-1}~{ m s}^{-1})$	$40 \pm 2 (35 \pm 10)$	10 (19)	11 (61%), 1.3 (39%) (16)	$63 \pm 12 \ (67 \pm 9)$	19 (12)	8.8 (38%), 2.1 (62%) (14)
•	$k_{\rm entry} \; (\mu {\rm M}^{-1} \; {\rm s}^{-1}) K_{\rm entry} \; ({\rm M}^{-1}) k_{\rm NO} \; (\mu {\rm M}^{-1} \; {\rm s}^{-1}) k_{\rm NOD} \; (\mu {\rm M}^{-1} \; {\rm s}^{-1}) \qquad k_{\rm O_2} \; (\mu {\rm M}^{-1} \; {\rm s}^{-1})$	50^{b}	42 (26)			23 (11)	9.4
	$k'_{\rm NO}~(\mu{\rm M}^{-1}~{\rm s}^{-1})$	31	18	23 (67%), 3.1 (33%)	89	21	8.3 (46%), 2.3 (54%)
	$K_{\rm entry}~({ m M}^{-1})$	2.0 ± 0.2	0.47	1.2, 0.086	8.4 ± 2.0	1.4	2.6, 0.13
	$k'_{\rm entry} \ (\mu { m M}^{-1} \ { m s}^{-1})$	33 ± 4	5.5	13, 2.9	42 ± 13	9.2	15, 1.6
,	$k'_{\rm CO}~(\mu{ m M}^{-1}~{ m s}^{-1})$	$5.2 \pm 0.5 \ (2.9 \pm 1.2)$	4.0 (3.7)	4.6 (50%), 0.85 (50%) (2.1)	$11 \pm 3.3 \ (6.8 \pm 1.5)$	7.8 (7.0)	4.5 (30%), 1.0 (70%) (5.9)
ò	$k_{\rm escape}$ (μs^{-1})	17 ± 2	12	11, 34	5.3 ± 0.4	6.5	5.8, 13
ì	$k_{\mathrm{bond}} \ (\mu \mathrm{s}^{-1})$	3.1 ± 0.4	31	17, 48 6.2, 14 11, 34	1.9 ± 0.3	37	2.4, 23
	$k_{\mathrm{gem}}~(\mu\mathrm{s}^{-1})$	20 ± 1	43	17, 48	7.1 ± 0.6	43	8.2, 36
				0.37 (56%), 0.29 (44%) (total = 0.67)			0.30 (44%), 0.65 (56%) (total = 0.95)
	subunit	lphaWT	$lpha$ Phe $(\mathrm{E}11)$	$a { m Trp}({ m E}11)$	$ ho_{ m WT}$	etaPhe $(E11)$	$ ho ext{Trp}(ext{E11})$
		dx	.doi.	org/10.	1021	/bi20	00923k1

phase are shown in parentheses. Parameter values were rounded to two significant digits, in the case of equilibrium constants after ratios were calculated. Kentry was calculated using the formula Kentry = for isolated WT α and β chains were taken from ref 31. In the k'_{CO} and k'_{O_2} columns, the values in parentheses represent the parameters for subunits within tetramers taken from ^αWT α and β subunits have valine at the E11 position. In cases where multiple bimolecular binding phases were observed, parameters for both processes are presented and relative contributions of each $^{b}k'_{\rm NOD}$ for WT ref 43. The $k_{
m NOD}$ values in parentheses are for lphaPhe(E11) and etaPhe(E11) mutants within tetramers and were taken from ref 34. Conditions were the same as those described for Figure 1. ^t 'kescape. K'CO and K'O2

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increased k_{bond} values. To simplify the analysis, we assumed that the two Trp(E11) phases for each subunit represent distinct noninterconverting conformations, fitted the time courses to two independent exponential expressions, and analyzed each phase in terms of eqs 1-3. In $\alpha Trp(E11)$, k_{bond} increases 2–5-fold; in $\beta \text{Trp}(\text{E}11)$, k_{bond} increases 10-fold for the larger phase, whereas the parameters for the smaller phase are similar to those for the WT β subunit. Variations in $k_{\rm escape}$ for the E11 mutants are small, ranging from ~2-fold increases to 50% decreases (Table 3).

The effects of the E11 mutations are caused by the loss of steric hindrance by the $\gamma 2$ methyl group of Val(E11) and confinement of the ligand close to the iron atom by the large aromatic side chains, both of which increase k_{bond} . Crystal structures of the Mb $\text{Trp}(\text{E}11)^{15}$ and $\text{Phe}(\text{E}11)^{44}$ mutants verify this interpretation and show both loss of steric hindrance near the Fe atom and a decrease in distal pocket volume compared to those of WT Mb. In both HbA and Mb, these effects result in greater rates and extents of geminate recombination in the E11 mutants.

The rate constants for bimolecular CO binding to the HbA Phe(E11) mutants are not dramatically different from the WT values, but 2-4-fold decreases in k'_{NO} and k'_{O} , are observed. Release of steric hindrance at the Fe atom as a result of the Val(E11) to Phe substitution increases k_{bond} , but this rate enhancement is offset by a reduction in the capture volume for ligands in the back of the distal cavity where the phenyl ring resides. These effects roughly compensate each other, accounting for the lack of change in k'_{CO} . However, in the case of NO binding, the rate-limiting step is ligand entry, the rate of which decreases in the presence of Phe(E11) due to the loss of distal pocket volume.

Both $\alpha Trp(E11)$ and $\beta Trp(E11)$ mutants reveal biphasic bimolecular ligand binding. In $\alpha Trp(E11)$, the fast phase parameters for bimolecular CO, O2, and NO binding are identical to those for the α Phe(E11) mutant, suggesting that the Trp residue has a conformation similar to that of the α Phe(E11) side chain. The slower binding phase probably represents an $\alpha \text{Trp}(\text{E}11)$ conformer in which the indole ring is filling more of the distal pocket cavity, greatly inhibiting entry of ligand into the active site. Similar conclusions can be drawn for the slow phase of bimolecular ligand binding to β Trp(E11). The values of k'_{NOD} for the oxygenated complexes of α Phe(E11) and α Trp(E11) mutants are similar to those for the WT subunits, suggesting that the more open conformations are preferred when O₂ is bound. However, significant decreases (3-5-fold) in k'_{NOD} are observed for the isolated or tetrameric β Phe(E11) and β Trp(E11) subunits.

Geminate and Bimolecular Ligand Binding to G8 Mutants. Phe and Trp substitutions at the G8 position do not produce significant changes in the rate and extent of slow, nanosecond geminate CO recombination from the C state in isolated Hb subunits (Figure 2D and Table 4). In α subunits, k_{bond} decreased only marginally for the Phe(G8) and Trp(G8) mutants, and similar 30% decreases were observed for k_{escape} . β Phe(G8) exhibited the largest change in F_{gem} , which decreased 2-fold because of a decrease in k_{bond} from 1.9 to $0.6~\mu \mathrm{s}^{-1}$ for the WT and mutant subunits. Both k_{bond} and k_{escape} decreased ~2-fold, for β Trp(G8), leaving F_{gem} unchanged. These latter changes suggest that the distal pocket of the β subunits is slightly expanded by the Trp(G8) indole side chain, allowing the ligand to linger in the distal pocket without

Table 4. Rate Constants for Bimolecular CO, O_2 , and NO Binding, NO Dioxygenation, and CO Geminate Recombination to Isolated α and β G8 Mutant Subunits^{α}

Konta Konta	subunit $F_{\rm gem}$ (μs^{-1}) (μs^{-1}) (μs^{-1}) $k_{\rm escape}$ (μs^{-1}) $k_{\rm CO}$ $(\mu M^{-1} s^{-1})$ $k_{\rm CO}$ $(\mu M^{-1} s^{-1})$ $k_{\rm NO}$ $(\mu M^{-1} s^{-1})$ $k_{\rm NO}$ $(\mu M^{-1} s^{-1})$ $k_{\rm OD}$ $(\mu M^{-1} s^{-1})$ $k_{\rm OD}$ $(\mu M^{-1} s^{-1})$	20 ± 1 3.1 ± 0.4 17 ± 2 5.2 ± 0.5 (2.9 ± 1.2) 33 ± 4 2.0 ± 0.2 31 50^b 40 ± 2 (35 ± 10)	13 2.0 11 3.6 (2.3) 24 2.1 40 55 30 (22)	14 1.7 12 2.0 (1.5) 16 1.3 30 74 19 (13)	7.1 ± 0.6 1.9 ± 0.3 5.3 ± 0.4 11 ± 3.3 (6.8 ± 1.5) 42 ± 1.3 8.4 ± 2.0 68 50^b 63 ± 12 (67 ± 9)	5.2 0.60 4.6 2.2 (1.0) 19 4.2 41 75 31 (19)	3.8 0.97 2.8 1.1 (0.42) 4.3 1.5 1.0 15 9.6 (4.2)
	$k_{\mathrm{gem}}~(\mu\mathrm{s}^{-1})$	20 ± 1	13	14	7.1 ± 0.6	5.2	3.8
	$F_{ m gem}$	0.16 ± 0.01	0.15	0.12	0.26 ± 0.02	0.12	0.26
	subunit	α WT	lphaPhe(G8)	m lpha Trp(G8)	$ ho_{ m WT}$	$\beta ext{Phe}(G8)$	$\beta \text{Trp}(G8)$

WT α and β subunits both have leucine at the G8 position. All values, calculated or measured, were rounded to two significant digits. $K_{
m entry}$ was calculated with the formula $K_{
m entry} = K_{
m entry}/K_{
m escape}$. $K_{
m CO}$ and for WT α and β chains were taken from ref 31. The values in parentheses in the k_{CO} and k_{O_2} columns represent the parameters for subunits within tetramers taken from ref 43. Conditions were the e as those described for Figure 1. $^{b}k_{\text{NOD}}$ values for α and β WT subunits within HbA tetramers.

same as those described

Biochemistry

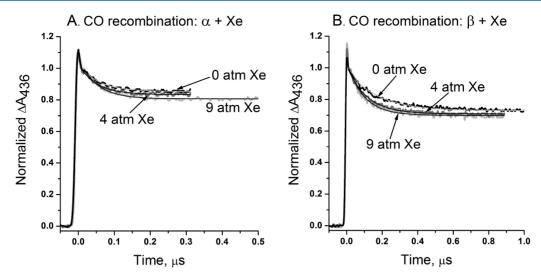


Figure 3. CO geminate rebinding to isolated α (A) and β (B) native subunits in the presence or absence of Xe. Kinetic measurements were conducted in a high-pressure cuvette (see Materials and Methods and ref 26) in 1 atm of CO with 0, 4, and 9 atm of Xe. The absorbance changes were normalized for more direct comparisons. The smooth lines represent best fits to a single-exponential expression for the slow phase. Experimental conditions were the same as those described for Figure 1.

Table 5. Effects of Xe on the Kinetic Parameters for Bimolecular CO and O₂ Binding and CO Geminate Recombination to Isolated α and β Subunits^a

subunit	$F_{ m gem}$	$k_{\rm gem}~(\mu {\rm s}^{-1})$	$k_{\rm bond}~(\mu {\rm s}^{-1})$	$k_{\rm escape}~(\mu {\rm s}^{-1})$	$k'_{\rm CO}~(\mu{\rm M}^{-1}~{\rm s}^{-1})$	$k'_{\rm entry}~(\mu\mathrm{M}^{-1}~\mathrm{s}^{-1})$	$K_{\rm entry}~({ m M}^{-1})$	$k'_{\rm O_2}~(\mu{\rm M}^{-1}~{\rm s}^{-1})$
α + 0 atm of Xe	0.16 ± 0.01	20 ± 1	3.1 ± 0.4	17 ± 2	5.8 ± 0.2	37 ± 3	2.2 ± 0.3	45 ± 3
α + 4 atm of Xe	0.19 ± 0.01	19 ± 1	3.7 ± 0.5	16 ± 2	6.1 ± 0.2	31 ± 2	2.0 ± 0.3	42 ± 3
α + 9 atm of Xe	0.24 ± 0.01	20 ± 1	4.8 ± 0.5	15 ± 1	6.0 ± 0.1	25 ± 2	1.7 ± 0.1	41
β + 0 atm of Xe	0.26 ± 0.02	7.1 ± 0.6	1.9 ± 0.3	5.3 ± 0.4	15 ± 0.2	56 ± 5	11 ± 1	74 ± 2
β + 4 atm of Xe	0.29 ± 0.02	8.8 ± 1.2	2.5 ± 0.6	6.3 ± 1.8	13 ± 0.2	45 ± 4	7.1 ± 2.1	65 ± 1
β + 9 atm of Xe	0.30 ± 0.02	11 ± 1	3.2 ± 0.6	7.4 ± 2.1	11 ± 0.3	36 ± 3	4.9 ± 1.4	59

"Average parameter values \pm standard deviations are presented when three separate experiments (on different days with different protein samples) had been performed. Parameter values were rounded to two significant digits after all calculations. K_{entry} was calculated with the formula $K_{\text{entry}} = k'_{\text{entry}}/k_{\text{escape}}$. Experimental conditions were the same as those described for Figure 1. In this table, only the results for O_2 and CO bimolecular binding conducted in the high-pressure stainless steel cuvette were used to calculate the mean values and standard deviations for k'_{entry} , k'_{O_2} , and k'_{CO} . The fitted parameters for 0 atm of Xe are not statistically different from those for WT α and β subunits reported in Tables 1–4 and ref 31.

Table 6. Effects of Xe on the Kinetic Parameters for Geminate CO Recombination to R-State HbA Tetramers and on the Bimolecular Association Rate Constants for CO Binding to T-State DeoxyHbA^a

subunit	$F_{ m gem}$	$F(\alpha+\beta)$	$k_{\mathrm{gem}}~(\mu\mathrm{s}^{-1})$	$k_{\mathrm{bond}}~(\mu\mathrm{s}^{-1})$	$k_{\rm escape}~(\mu { m s}^{-1})$	$k'_{\mathrm{TCO}}(\alpha+\beta)~(\mu\mathrm{M}^{-1}~\mathrm{s}^{-1})$
Hb α + 0 atm of Xe	0.28 ± 0.06	0.31 ± 0.03	34 ± 3	9.4 ± 1.3	24 ± 4	0.24 ± 0.01
Hb β + 0 atm of Xe	0.33 ± 0.03		10 ± 2	3.5 ± 0.8	7 ± 1	
Hb α + 4 atm of Xe	0.43 ± 0.07	0.37 ± 0.04	30 ± 3	13 ± 2	17 ± 2	0.22
Hb β + 4 atm of Xe	0.32 ± 0.06		9.9 ± 3	3.2 ± 1.1	6.7 ± 1.7	
Hb α + 9 atm of Xe	0.44 ± 0.05	0.42 ± 0.02	33 ± 3	15 ± 1	18 ± 3	0.24
Hb β + 9 atm of Xe	0.38 ± 0.04		11 ± 1	4 ± 1	6.5 ± 0.8	

"When three or more independent experiments had been conducted, the measured or calculated parameters are presented as the mean value \pm the standard deviation. Parameter values were rounded to two significant digits. $F(\alpha+\beta)$ represents the average of $F_{\rm gem}$ for the α and β subunits within the tetramer, and $F_{\rm gem}$ represents the total fraction of geminate recombination observed for the Hb sample. The $F_{\rm gem}$ values for the individual subunits were calculated as $F_{\rm gem}(\alpha)$ or $F_{\rm gem}(\alpha)$ is the observed amplitude of geminate rebinding assigned to an individual subunit within a tetramer. The experimental conditions were the same as those described for Figure 1.

affecting the probability of recombination or escape. In contrast, the Phe(G8) and Trp(G8) substitutions increase both $k_{\rm bond}$ and $k_{\rm escape}$ in Mb, which is consistent with a decreased distal pocket volume.³

The HbA G8 mutants do cause measurable changes in the overall bimolecular kinetic parameters (Table 4). Decreases in $k'_{\rm CO}$ and $k'_{\rm O_2}$ with an increase in the size of the G8 residue are observed for both subunits. However, the $k'_{\rm NO}$, $k'_{\rm NOD}$, and

calculated k'_{entry} values remain relatively unchanged (\leq 2-fold). Only the β Trp(G8) mutant shows large (\geq 5-fold) decreases in the bimolecular rate constants for the binding of all three gases, ligand entry, and NO dioxygenation. Thus, only in β subunits does the Leu(G8) to Trp mutation cause a functionally measurable decrease in the ligand capture volume.

Effects of Xe Pressure on Ligand Binding to HbA and Its Subunits. Time courses for geminate CO recombination

to isolated WT Hb subunits under high pressures of Xe gas are shown in Figure 3, and the resultant kinetic parameters are listed in Table 5. Nonspecific effects of pressure on binding of ligand to globins have been ruled out by previous work. 45,26 Small increases in $F_{\rm gem}$ are detected for isolated native $\alpha {\rm CO}$ subunits when the Xe partial pressure is increased to 9 atm and are due to an ~50% increase in $k_{\rm bond}$ from 3.1 to 4.8 μs^{-1} . The values of k_{escape} , k'_{O_2} , and k'_{CO} for isolated α subunits show no statistically significant changes. Time courses for geminate CO rebinding to β subunits at high Xe pressures are shown in Figure 3B. Again, F_{gem} increases slightly because of a small increase in k_{bond} , and in this case, the bimolecular rate constants for ligand binding do decrease ~20% at high pressures of Xe; however, this change is probably not statistically significant (Table 5). Thus, filling internal globin cavities with Xe has very little effect on either geminate or bimolecular ligand binding in the α and β subunits in contrast to the space-filling Phe and Trp mutations at the B10 and E11 helical positions.

The kinetic parameters for internal CO rebinding within intact HbA tetramers are also little affected by Xe pressure, even though the fraction of geminate recombination is higher under all conditions (Table 6). Again, a small increase in $F_{\rm gem}$ is observed for α subunits at 9 atm of Xe because of a 60% increase in $k_{\rm bond}$ with little or no change in $k_{\rm escape}$. However, the key result in Table 6 is that Xe binding has little or no effect on ligand binding to tetramers or monomers.

To assess the influence of Xe atoms on ligand binding to T-state HbA, we measured CO rebinding to Hb tetramers after full photolysis, which allows the protein to switch to the low-affinity T-state quaternary conformation. The time courses at varying Xe pressures are virtually indistinguishable from those in the absence of Xe (not shown). The rate constants for CO binding to T-state HbA ($k'_{\rm TCO}$) were determined by analyzing the slow, accelerating phases of each full photolysis trace. No changes in $k'_{\rm TCO}$ were detected upon addition of Xe (Table 6).

DISCUSSION

Structural Interpretations. The crystal structures of recombinant HbA variants containing B10, E11, and G8 mutations were determined previously following procedures described in ref 46, and the coordinates were deposited in the Protein Data Bank by E. A. Bruker in 2002 (entries 101I-101P). The most relevant structures for interpreting our kinetic studies include the deoxy forms of $\alpha_1(Met(NA1)/Trp(B10))$ - $(Gly)_3-\alpha_2(Trp(B10)):\beta(Met(NA1))$ (101N), $\alpha_1(Met(NA1))$ Phe(B10)/Gln(E7))-Gly- α_2 -(Phe(B10)/Gln(E7)): β (Met-(NA1)/Trp(G8)) (1O1J), and α_1 (Met(NA1)/Phe(B10)/Gln-(E7))-(Gly)₃- α_2 (Phe(B10)/Gln(E7)): β (Met(NA1)/Trp-(E11)) (101M), and the cyanomet form of $\alpha_1(\text{Met}(\text{NA1})/$ Phe(B10)/Gln(E7))-Gly- α_2 -(Phe(B10)/Gln(E7)): β (Met(NA1)/ Trp(G8)) (101I). One to three Gly residues were introduced between the α_1 and α_2 subunits in these recombinant Hbs to prevent dissociation into dimers as part of an effort to design a safe and effective Hb-based oxygen carrier. 33 The Met(NA1) mutation is not present in the α_2 subunit domain because the α_1 and α_2 subunits were expressed as a single di- α polypeptide.

The structures of the distal pockets of deoxy- α Trp(B10), deoxy- α Phe(B10)/Gln(E7), deoxy- β Trp(E11), and cyanomet- β Trp(G8) subunits in WT/mutant hybrid tetramers of recombinant HbA are presented in Figure 4. Panels A and B show clearly that the immediate vicinity of the Fe atom is restricted by large aromatic side chains at the B10 position in α

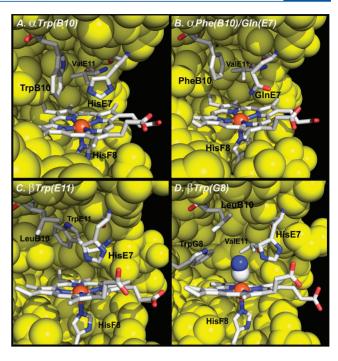


Figure 4. Structures of the key B10, E11, and G8 mutants of HbA. The models of the mutant distal pockets were constructed from the following structures: (A) deoxyHb $\alpha(\text{Trp}(B10)):\beta(WT)$ (1O1N), (B) deoxyHb $\alpha(\text{Phe}(B10)/\text{Gln}(E7)):\beta(\text{Trp}(G8))$ (1O1J), (C) deoxyHb $\alpha(\text{Phe}(B10)/\text{Gln}(E7)):\beta(\text{Trp}(E11))$ (1O1M), and (D) cyanometHb $\alpha(\text{Phe}(B10)/\text{Gln}(E7)):\beta(\text{Trp}(G8))$ (1011). In addition to the listed mutations, each structure contains Gly links between α chains and Met(NA1) replacements in both subunits. The key residues of the heme pocket are labeled and shown as sticks. The atoms of the key amino acids are colored as follows: white, carbon; blue, nitrogen; red, oxygen; yellow, atoms of other amino acids lining the binding site; and orange, heme Fe. In panel D, white and blue spheres represent the CN ligand bound to Fe3+. In each panel, part of the distal pocket was removed to show the internal structure. All molecular drawings (Figures 4-6 and 8) were prepared using PyMol (PyMol Molecular Graphics System, version 1.2r3pre, Schrödinger, LLC, New York, NY).

chains. The distances from Fe to the closest edges of the $\alpha \text{Phe}(\text{B10})$ and $\alpha \text{Trp}(\text{B10})$ side chains are 5.5 and 4.8 Å, respectively. In native α subunits, the distance from Fe to the $\text{C}\delta 2$ methyl of Leu(B10) is 7.4 Å. In WT α subunits, the Leu(B10) and His(E7) side chains are sufficiently far from the Fe and sufficiently flexible to allow up to 30% geminate CO recombination. In contrast, the aromatic rings of $\alpha \text{Phe}(\text{B10})$ and Trp(B10) side chains sterically hinder the formation of the first geminate state (state B in Scheme 1), push the ligands back to the iron atom (ultrafast phase) or out of the distal pocket toward either solvent or the protein interior, and almost completely abolish nanosecond geminate rebinding. The conformations of the $\alpha \text{Phe}(\text{B10})$ and Trp(B10) side chains shown in Figure 4 are almost identical to those of Phe(B10) and Trp(B10) in Mb mutants. 15,44

When Trp is substituted for Val(E11) in β subunits, two significant structural changes occur (Figure 4C). First, the indole side chain fills the accessible space in the back of the distal pocket, markedly reducing the capture volume in the distal pocket. Second, loss of the γ 2 methyl group of Val(E11) removes steric restriction adjacent to the Fe atom, enhancing the rate of internal bond formation. These structural features explain the dramatic increases in the fractions and rates of CO

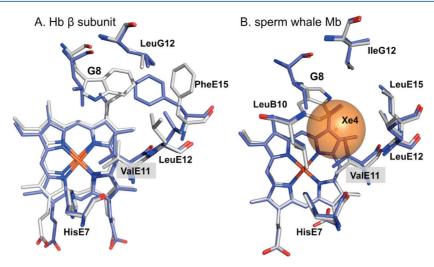


Figure 5. Distal pocket structures of (A) oxy- β (2DN1) and cyanomet- β (Trp(G8)) (101I) HbA and (B) WT (2MBW) and Trp(G8) (2OHB) sperm whale aquometMb. The atoms are colored as in Figure 4, except that the carbons are colored white in the mutant structures and blue in the WT structures for both β subunits and Mb. The large brown sphere in panel B represents the Xe4 cavity in sperm whale Mb.

geminate recombination in all four Hb Phe(E11) and Trp(E11) mutants and in the corresponding Mb E11 variants. 15,44

Neither α - nor $\beta Trp(G8)$ mutations cause any significant effects on CO geminate recombination, suggesting that this amino acid is not part of a ligand migration pathway toward cavities in the protein interior. In contrast, the Mb Ile(G8) to Trp mutation does cause an almost complete loss of the slow microsecond phase of O2 geminate recombination that is associated with movement of ligand into and return of ligand from the Xe1 and Xe4 sites.^{3,15} Structural differences between β Trp(G8) subunits and Mb Trp(G8) are shown in Figure 5. The structure of the WT β HbO₂ distal pocket (blue) is superimposed on the model for the cyanomet- β Trp(G8) mutant (white) in panel A. In WT β subunits, the side chain of Phe(E15) is close to the pyrrole B ring, is parallel to the heme plane, and fills an area spatially similar to the Xe4 cavity in Mb (brown sphere above the heme plane in Figure 5B). In the $\beta Trp(G8)$ mutant, the Phe(E15) side chain has rotated ~90° away from the heme to accommodate the bulky G8 indole group, which occupies part of the same space that the E15 phenyl ring fills in the β WT distal pocket. Thus, the net change in accessible distal pocket volume caused by the Leu(G8) to Trp mutation is relatively small, and there is no direct route to the protein interior in either βWT or the Trp(G8) variant.

In Mb, the Trp(G8) side chain does not displace any distal pocket side chains and, instead, assumes a conformation in which the indole ring is located above the edge of the pyrrole B ring, reduces the size of the capture volume, and blocks access to the Xe4 cavity, which is above it in Figure 5B. Thus, in Mb, Trp(G8) accelerates bond formation by sequestering ligands closer to the Fe atom and blocking ligand access to the Xe1 and Xe4 binding sites.^{3,15}

In the deoxy- β Trp(G8) distal pocket (Figure 6), the conformation of the G8 indole side chain is similar to that in the cyanomet form; however, a water molecule is coordinated to the pyrrole nitrogen of the indole ring and the carbonyl O atoms of Gly(B6) and Gly(E8). The distances between the water and these atoms of Trp(G8), Gly(B6), and Gly(E8) (3.2, 2.8, and 3.1 Å, respectively) indicate the formation of strong hydrogen bonds. As a result, the distal capture volume in the

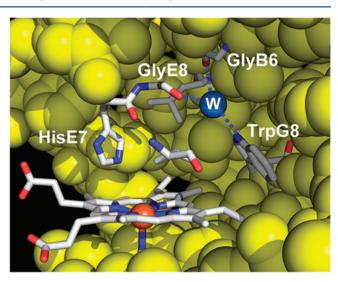


Figure 6. Structure of the deoxy-βTrp(G8) distal pocket. This model was constructed from the structure of deoxyHb α_1 (Met(NA1)/Phe(B10)/Gln(E7))-Gly- α_2 -(Phe(B10)/Gln(E7)):β(Met(NA1)/Trp(G8)) (1O1J) using PyMol. The blue sphere corresponds to distal pocket water molecule; blue dotted lines represent hydrogen bonds. The atoms are colored as in Figure 4. Part of the distal pocket was removed to show the internal structure.

equilibrium form of deoxy- β Trp(G8) is significantly smaller than that in WT β subunits, resulting in a 4-fold decrease in k'_{entry} (Table 4). However, this water molecule is not present in the initial liganded state of β Trp(G8) and has no effect on geminate rebinding. This internalization of a water molecule in the equilibrium structure of the mutant deoxy- β Trp(G8) subunit explains the apparent discrepancy between the 4-fold decreased k'_{entry} and the lack of an effect on geminate recombination, which occurs before water can enter the distal pocket (Table 4).

Ligand Trapping in the Distal Pocket. The equilibrium constant for ligand entry can be computed with the formula $K_{\text{entry}} = k'_{\text{entry}}/k_{\text{escape}}$ to quantify the noncovalent ligand capture efficiency of the distal pocket. This constant depends primarily on the volume available to ligands in or near the active site,

which serves to retain the captured ligand until it binds to the heme Fe. For native α and β subunits, these values are 2.0 ± 0.2 and 8.4 ± 2.0 M⁻¹, respectively, based on our data and previous work. The presence of a distal pocket water hydrogen bonded to His(E7) in WT α subunits and the lack of flexibility of Leu(B10) contribute to its smaller $K_{\rm entry}$ value compared to that of β subunits (Figure 6 and refs 34, 48, and 49). Significant decreases in $K_{\rm entry}$ occur for the Phe and Trp(B10) and Trp(E11) mutants, with the Trp replacements showing 4 to 1000-fold decreases, depending on which geminate phase is analyzed (Tables 2 and 3, the bars in Figure 7 represent the

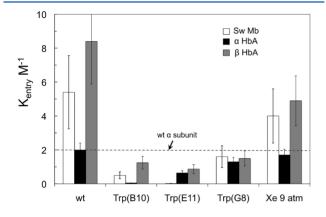


Figure 7. Effects of Trp mutations and Xe on the equilibrium constant for noncovalent ligand capture ($K_{\rm entry}$). The errors for the WT values were computed for at least three independent determinations of $K_{\rm entry}$, including previous data for both O₂ and CO binding, and are assumed to apply to the mutants. ^{15,31,37} In this figure, the values for the mutant subunits were computed as an amplitude-weighted average of $K_{\rm entry}$ for the fast and slow geminate phases listed in Tables 2–4 (*i.e.*, $K_{\rm entry,ave}$ for (TrpB10) = 0.62 × 2.0 + 0.38 × 0.013 = 1.2 M⁻¹). This value under represents the dramatic decreases in $K_{\rm entry}$ for the closed and restricted conformations but does reflect the apparent, average accessibility of the variants to ligands. The dashed line represents the value of 2.0 M⁻¹ for WT α subunits.

weighted average K_{entry}). The Trp(G8) replacement has no effect on K_{entry} in α subunits (Table 4 and Figure 7), whereas this mutation causes 3–4-fold decreases in K_{entry} in Mb and β subunits. The decrease in capture volume in Mb is due to the orientation of the G8 indole side chain over the heme group, which blocks access to the Xe4 cavity (Figure 5B). The decrease in K_{entry} for deoxy- β Trp(G8) is due primarily to the presence of a water molecule hydrogen bonded to the indole N atom (Figure 6).

The results in Figure 7 also show that the addition of Xe has little effect on the accessibility of the noncovalent capture volume in either Mb or the α subunits of HbA. The only statistically significant effect is observed for β subunits, where K_{entry} decreased \leq 2-fold in the presence of 9 atm of Xe (Table 5 and Figure 7). This decrease suggests that a small fraction of Xe atoms can penetrate into the β distal pocket and reduce the volume of the noncovalent binding site. In contrast, the α distal pocket appears to be too small to accommodate a Xe atom.

Ultrafast Phase of Geminate Recombination and **Secondary Sites in HbA.** Geminate O₂ recombination in Mb is multiexponential with a fast, nanosecond phase reflecting ligand association from the distal pocket and adjacent Xe4 cavity and slower, microsecond phases representing ligand rebinding from the more distant Xe1 site. 3,4,15,50 Sottini et al. 28,29 reported multiple geminate phases in human HbA similar to those shown in Figures 1-3. ^{28,29} On the basis of the mechanism for geminate O₂. recombination to Mb, these authors assigned the slower phase to ligand rebinding from remote sites, distinct from the distal pocket. The faster phase was assigned to ligand recombination within the distal pocket. In contrast, our results suggest that there are no remote, secondary docking sites in HbA analogous to those seen in Mb. We assign the slower nanosecond phase for CO geminate recombination in HbA to ligand rebinding within the distal pocket or sites closely adjacent to it. The ultrafast phase was assigned to rebinding from an initial geminate state, where the ligand is transiently very close to the Fe atom (Scheme 1, state B). Our interpretation is based on the following arguments.

First, the slower phase for CO geminate recombination still occurs with the same amplitude in the presence of high

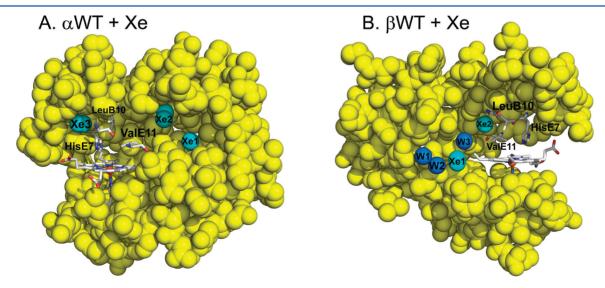


Figure 8. Structures of the deoxy α (A) and β (B) subunits with bound Xe atoms. The structures were constructed from PDB entry 2W6V.³⁰ The key residues of the heme pocket are labeled and shown as sticks. The atoms of key amino acids are represented as sticks, colored as in Figure 4. Cyan spheres represent Xe, and blue spheres indicate positions of water molecules. In the β subunit, water molecules are shown to emphasize possible ligand migration pathways. In panel B, W1 and W2 are external waters, whereas W3 and the Xe atoms are buried inside the protein.

pressures of Xe gas in both isolated subunits and tetrameric HbA (Figure 3). In contrast, high Xe pressures eliminate the slow O2 recombination phase in Mb due to occupation of secondary Xe4 and Xe1 sites. Second, the slower CO recombination phase in the HbA subunits occurs with a halftime of $\sim 50-100$ ns, which is similar to the faster O_2 recombination phase seen in WT Mb and contrasts with the slow CO and O2 geminate phases in WT Mb, which have halftimes of $\approx 1-2 \mu s$. Third, the pathways leading to the proposed Xe binding sites in HbA (refs 28-30 and Figure 8) should be sterically hindered by the space-filling E11 mutations, especially by the Trp replacements, which would be expected to decrease the fraction of the slow geminate recombination and increase the fraction of the fast phase. In our experiments, however, the slow phases for CO geminate recombination do not disappear and the ultrafast phases are not observed at all in the geminate time courses for the α and β E11 mutants (Figure 2C). The absence of the ultrafast phase for the internal CO rebinding in E11 mutants of HbA suggests that the naturally occurring Val(E11) γ2 methyl group is the structural barrier between states B and C in Scheme 1.

Ligand Migration via Alternative Routes in HbA. Figure 8 shows the crystal structures of the Hb α and β subunits with bound Xe atoms as determined by Savino et al. These authors proposed that the cavities occupied by Xe atoms transiently open and form a channel within the protein matrix that guides migration of ligand from solvent to the distal pocket and vice versa. One of these tunnels is more pronounced in β subunits, contains a string of well-ordered waters and Xe atoms, and appears to identify a potential migratory pathway that is analogous to the apolar tunnel in *C. lacteus* mini-Hb. 25,26

This proposed tunnel in β subunits can be blocked by mutating Val(E11) to Trp (see Figures 4C and 8B). In β Trp(E11), the indole ring of the Trp is located in the space above the B ring of the porphyrin. The Xe2 atom occupies that same position in the structure of β WT at high gas pressures (Figure 8B). The dramatic increase in the fraction of recombination due to the presence of Trp(E11) (Table 3 and Figure 2C) could be interpreted as supporting the alternative migration route through the Xe2 cavity. However, the calculated rates of escape do not decrease in the Trp(E11) mutants but instead increase 2-fold (Table 3). The observed increases in the fractions of recombination in the Hb Trp(E11) mutants are due almost exclusively to markedly larger rates of ligand-Fe bond formation (k_{bond} in Table 3) and not to slow rates of escape. More importantly, Xe binding does not show any significant inhibition of ligand entry or escape (Tables 5 and 6).

The absence of effects of Xe on ligand recombination in HbA is not a surprise. In Mb, increasing the Xe partial pressure does not change the fraction of CO recombination, which is inherently very small (~0.05). In the case of geminate O₂ rebinding to Mb, the total amount of recombination also stays unchanged upon addition of up to ~10 atm of Xe; however, the slow microsecond phase of the geminate reaction does disappear because ligands can no longer enter the Xe1 and Xe4 sites. The lack of an effect on the overall fraction of geminate recombination suggests strongly that, even though ligands do visit the Xe binding sites in the Mb protein matrix, these cavities do not serve as ports of entry or escape. In Scapharca inaequivalvis HbI dimers, Xe binding also has no effect on the fraction of ligand escape even though MD

simulations suggest that ligands migrate into solvent through Xe binding cavities and. ²³ Only in the mini-Hb from *C. lacteus* does Xe binding significantly increase the fraction of geminate CO recombination (increasing $F_{\rm gem}$ from 0.05 to ~0.30) and decrease the rate of entry of ligand into the protein, confirming that its apolar tunnel is the major route for ligand entry and escape. ²⁶

A key remaining question is whether the His(E7) pathway also applies in T-state deoxyhemoglobin or is affected by the R to T quaternary transition. At present, there is little evidence suggesting that the pathways for movement of ligand into the active sites of the α and β subunits are altered markedly by the quaternary transition. First, the conformations of the amino acids in and near the His(E7) gate are not dramatically altered upon comparison of high-resolution deoxyHbA and oxyHbA structures, except for the inward movement of the distal histidine when the ligand is not present and some movement of the E helix across the distal face of the heme.⁴⁹ Second, this gate is exposed to solvent and far from both the $\alpha 1 - \beta 1$ and $\alpha 1 - \beta 2$ interfaces.⁴⁹ Third, Cassoly and Gibson⁵¹ have shown that the bimolecular rate constant for NO association is roughly the same for the first and last steps in ligand binding, and we have observed the same result. 52 Because of its high reactivity, the limiting step for NO binding is diffusion into the active site. Thus, there is very little change in k'_{entry} between the R and T states, and probably the same pathway applies in each case. However, a more detailed analysis of bimolecular rate constants for the T-state forms of mutant Hb tetramers is needed to verify this conclusion.

CONCLUSION

Our results for space-filling mutations and Xe binding, combined with previous kinetic data for His(E7) mutations, 31,32 suggest strongly that the "baseball glove" model for ligand binding, originally proposed for sperm whale Mb,³ also applies to the α and β subunits of human HbA, either when isolated or present in R-state tetramers. Reducing the volume of the noncovalent binding site in both subunits with Phe or Trp(B10) and Trp(E11) mutations weakens the ability of hemoglobin to retain captured ligands for bond formation with the Fe atom. The small internal cavities, analogous to the Xe pockets in Mb, do not appear to host ligands, even transiently during nanosecond laser-flash photolysis experiments. In both subunits of HbA, the fast and slow phases of geminate recombination appear to represent rebinding from an initial transient geminate state formed immediately after ligand photodissociation and then from more interior regions in or very near the distal pocket, respectively. The system of Xe-binding sites observed in crystal structures of HbA does not appear to be part of a major pathway for ligand migration because filling these cavities with Xe gas does not significantly affect the rates of ligand escape or bimolecular entry.

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ABBREVIATIONS

Hb, hemoglobin; α and β , subunits of adult human hemoglobin (HbA), which can be isolated biochemically or present in intact heterotetramers or $\alpha\beta$ dimers; Mb, myoglobin; WT, wild-type; NOD, NO dioxygenation; $k_{\rm X}$, association rate constant for ligand X; $k_{\rm X}$, dissociation rate for ligand X; $F_{\rm gem}$, fraction of geminate recombination; $k_{\rm gem}$, rate of geminate recombination; $k_{\rm bond}$, rate of internal bond formation; $k_{\rm escape}$, rate of escape of ligand from the protein; $k'_{\rm entry}$, rate of entry of ligand into distal pocket; $K_{\rm entry}$, equilibrium constant for ligand capture in the noncovalent binding site calculated as $k'_{\rm entry}/k_{\rm escape}$; Nd:YAG, neodymium-doped yttrium aluminum garnet; rmsd, root-mean-square deviation.

REFERENCES

- (1) Schoenborn, B. P., Watson, H. C., and Kendrew, J. C. (1965) Binding of xenon to sperm whale myoglobin. *Nature* 207, 28–30.
- (2) Tilton, R. F. Jr., Kuntz, I. D. Jr., and Petsko, G. A. (1984) Cavities in proteins: Structure of a metmyoglobin-xenon complex solved to 1.9 Å. *Biochemistry* 23, 2849–2857.
- (3) Scott, E. E., Gibson, Q. H., and Olson, J. S. (2001) Mapping the pathways for O_2 entry into and exit from myoglobin. *J. Biol. Chem.* 276, 5177–5188.
- (4) Scott, E. E., and Gibson, Q. H. (1997) Ligand migration in sperm whale myoglobin. *Biochemistry* 36, 11909–11917.
- (5) Tetreau, C., Blouquit, Y., Novikov, E., Quiniou, E., and Lavalette, D. (2004) Competition with xenon elicits ligand migration and escape pathways in myoglobin. *Biophys. J.* 86, 435–447.
- (6) Radding, W., and Phillips, G. N. Jr. (2004) Kinetic proofreading by the cavity system of myoglobin: Protection from poisoning. *BioEssays* 26, 422–433.
- (7) Ostermann, A., Waschipky, R., Parak, F. G., and Nienhaus, G. U. (2000) Ligand binding and conformational motions in myoglobin. *Nature* 404, 205–208.
- (8) McNaughton, L., Hernandez, G., and LeMaster, D. M. (2003) Equilibrium O_2 distribution in the Zn^{2+} -protoporphyrin IX

deoxymyoglobin mimic: Application to oxygen migration pathway analysis. J. Am. Chem. Soc. 125, 3813–3820.

- (9) Case, D. A., and Karplus, M. (1979) Dynamics of ligand binding to heme proteins. *J. Mol. Biol.* 132, 343–368.
- (10) Elber, R., and Karplus, M. (1990) Enhanced sampling in molecular dynamics: Use of the time dependent Hartree approximation for simulation of carbon monoxide diffusion through myoglobin. *J. Am. Chem. Soc.* 112, 9161–9175.
- (11) Cohen, J., and Schulten, K. (2007) O_2 migration pathways are not conserved across proteins of a similar fold. *Biophys. J.* 93, 3591–3600
- (12) Cohen, J., Olsen, K. W., and Schulten, K. (2008) Finding gas migration pathways in proteins using implicit ligand sampling. *Methods Enzymol.* 437, 439–457.
- (13) Elber, R., and Gibson, Q. H. (2008) Toward quantitative simulations of carbon monoxide escape pathways in myoglobin. *J. Phys. Chem. B* 112, 6147–6154.
- (14) Ruscio, J.Z., Kumar, D., Shukla, M., Prisant, M. G., Murali, T. M., and Onufriev, A. V. (2008) Atomic level computational identification of ligand migration pathways between solvent and binding site in myoglobin. *Proc. Natl. Acad. Sci. U.S.A.* 105, 9204–9209
- (15) Olson, J. S., Soman, J., and Phillips, G. N. Jr. (2007) Ligand pathways in myoglobin: A review of Trp cavity mutations. *IUBMB Life* 59, 552–562.
- (16) Morikis, D., Champion, P. M., Springer, B. A., and Sligar, S. G. (1989) Resonance Raman investigations of site-directed mutants of myoglobin: Effects of distal histidine replacement. *Biochemistry* 28, 4791–4800.
- (17) Rohlfs, R. J., Mathews, A. J., Carver, T. E., Olson, J. S., Springer, B. A., Egeberg, K. D., and Sligar, S. G. (1990) The effects of amino acid substitution at position E7 (residue 64) on the kinetics of ligand binding to sperm whale myoglobin. *J. Biol. Chem.* 265, 3168–3176.
- (18) Yang, F., and Phillips, G. N. Jr. (1996) Crystal structures of CO-, deoxy- and met-myoglobins at various pH values. *J. Mol. Biol.* 256, 762–774.
- (19) Tian, W. D., Sage, J. T., and Champion, P. M. (1993) Investigations of ligand association and dissociation rates in the "open" and "closed" states of myoglobin. *J. Mol. Biol.* 233, 155–166.
- (20) Tian, W. D., Sage, J. T., Champion, P. M., Chien, E., and Sligar, S. G. (1996) Probing heme protein conformational equilibration rates with kinetic selection. *Biochemistry* 35, 3487–3502.
- (21) Lai, H. H., Li, T., Lyons, D. S., Phillips, G. N. Jr., Olson, J. S., and Gibson, Q. H. (1995) Phe-46(CD4) orients the distal histidine for hydrogen bonding to bound ligands in sperm whale myoglobin. *Proteins* 22, 322–339.
- (22) Nienhaus, K., Knapp, J. E., Palladino, P., Royer, W. E. Jr., and Nienhaus, G. U. (2007) Ligand migration and binding in the dimeric hemoglobin of *Scapharca inaequivalvis*. *Biochemistry* 46, 14018–14031.
- (23) Knapp, J. E., Pahl, R., Cohen, J., Nichols, J. C., Schulten, K., Gibson, Q. H., Srajer, V., and Royer, W. E. Jr. (2009) Ligand migration and cavities within *Scapharca* dimeric HbI: Studies by time-resolved crystallography, Xe binding, and computational analysis. *Structure* 17, 1494–1504.
- (24) Boechi, L., Marti, M. A., Milani, M., Bolognesi, M., Luque, F. J., and Estrin, D. A. (2008) Structural determinants of ligand migration in *Mycobacterium tuberculosis* truncated hemoglobin O. *Proteins* 73, 372–370
- (25) Salter, M. D., Nienhaus, K., Nienhaus, G. U., Dewilde, S., Moens, L., Pesce, A., Nardini, M., Bolognesi, M., and Olson, J. S. (2008) The apolar channel in *Cerebratulus lacteus* hemoglobin is the route for O₂ entry and exit. *J. Biol. Chem.* 283, 35689–35702.
- (26) Pesce, A., Nardini, M., Dewilde, S., Capece, L., Marti, M. A., Congia, S., Salter, M. D., Blouin, G. C., Estrin, D. A., Ascenzi, P., Moens, L., Bolognesi, M., and Olson, J. S. (2011) Ligand migration in

the apolar tunnel of *Cerebratulus lacteus* mini-hemoglobin. *J. Biol. Chem.* 286, 5347–5358.

- (27) Adachi, S., Park, S. Y., Tame, J. R., Shiro, Y., and Shibayama, N. (2003) Direct observation of photolysis-induced tertiary structural changes in hemoglobin. *Proc. Natl. Acad. Sci. U.S.A.* 100, 7039–7044.
- (28) Sottini, S., Abbruzzetti, S., Spyrakis, F., Bettati, S., Ronda, L., Mozzarelli, A., and Viappiani, C. (2005) Geminate rebinding in R-state hemoglobin: Kinetic and computational evidence for multiple hydrophobic pockets. *J. Am. Chem. Soc. 127*, 17427–17432.
- (29) Sottini, S., Abbruzzetti, S., Viappiani, C., Bettati, S., Ronda, L., and Mozzarelli, A. (2005) Evidence for two geminate rebinding states following laser photolysis of R state hemoglobin encapsulated in wet silica gels. *J. Phys. Chem. B* 109, 11411–11413.
- (30) Savino, C., Miele, A. E., Draghi, F., Johnson, K. A., Sciara, G., Brunori, M., and Vallone, B. (2009) Pattern of cavities in globins: The case of human hemoglobin. *Biopolymers 91*, 1097–1107.
- (31) Birukou, I., Schweers, R. L., and Olson, J. S. (2010) Distal histidine stabilizes bound O₂ and acts as a gate for ligand entry in both subunits of adult human hemoglobin. *J. Biol. Chem.* 285, 8840–8854.
- (32) Birukou, I., Soman, J., and Olson, J. S. (2011) Blocking the gate to ligand entry in human hemoglobin. *J. Biol. Chem.* 286, 10515–10529
- (33) Looker, D., Mathews, A. J., Neway, J. O., and Stetler, G. L. (1994) Expression of recombinant human hemoglobin in *Escherichia coli*. *Methods Enzymol*. 231, 364–374.
- (34) Eich, R. F., Li, T., Lemon, D. D., Doherty, D. H., Curry, S. R., Aitken, J. F., Mathews, A. J., Johnson, K. A., Smith, R. D., Phillips, G. N. Jr., and Olson, J. S. (1996) Mechanism of NO-induced oxidation of myoglobin and hemoglobin. *Biochemistry* 35, 6976–6983.
- (35) Olson, J. S., Foley, E. W., Rogge, C., Tsai, A. L., Doyle, M. P., and Lemon, D. D. (2004) NO scavenging and the hypertensive effect of hemoglobin-based blood substitutes. *Free Radical Biol. Med.* 36, 685–697.
- (36) Draghi, F., Miele, A. E., Travaglini-Allocatelli, C., Vallone, B., Brunori, M., Gibson, Q. H., and Olson, J. S. (2002) Controlling ligand binding in myoglobin by mutagenesis. *J. Biol. Chem.* 277, 7509–7519.
- (37) Olson, J. S., Rohlfs, R. J., and Gibson, Q. H. (1987) Ligand recombination to the α and β subunits of human hemoglobin. *J. Biol. Chem.* 262, 12930–12938.
- (38) Olson, J. S., and Phillips, G. N. Jr. (1996) Kinetic pathways and barriers for ligand binding to myoglobin. *J. Biol. Chem.* 271, 17593—17596
- (39) Jongeward, K. A., Magde, D., Taube, D. J., Marsters, J. C., Traylor, T. G., and Sharma, V. S. (1988) Picosecond and nanosecond geminate recombination of myoglobin with CO, O₂, NO, and isocyanides. *J. Am. Chem. Soc.* 110, 380–387.
- (40) Carver, T. E., Rohlfs, R. J., Olson, J. S., Gibson, Q. H., Blackmore, R. S., Springer, B. A., and Sligar, S. G. (1990) Analysis of the kinetic barriers for ligand binding to sperm whale myoglobin using site-directed mutagenesis and laser photolysis techniques. *J. Biol. Chem.* 265, 20007–20020.
- (41) Miele, A. E., Draghi, F., Arcovito, A., Bellelli, A., Brunori, M., Travaglini-Allocatelli, C., and Vallone, B. (2001) Control of heme reactivity by diffusion: Structural basis and functional characterization in hemoglobin mutants. *Biochemistry* 40, 14449–14458.
- (42) Maillett, D. H., Simplaceanu, V., Shen, T. J., Ho, N. T., Olson, J. S., and Ho, C. (2008) Interfacial and distal-heme pocket mutations exhibit additive effects on the structure and function of hemoglobin. *Biochemistry* 47, 10551–10563.
- (43) Maillett, D. H. (2003) Engineering Hemoglobins and Myoglobins for Efficient O₂ Transport. In *Biochemistry and Cell Biology*, Rice University, Houston.
- (44) Quillin, M. L., Li, T., Olson, J. S., Phillips, G. N. Jr., Dou, Y., Ikeda-Saito, M., Regan, R., Carlson, M., Gibson, Q. H., Li, H., et al.

- (1995) Structural and functional effects of apolar mutations of the distal valine in myoglobin. *J. Mol. Biol.* 245, 416–436.
- (45) Wiltrout, M. E., Giovannelli, J. L., Simplaceanu, V., Lukin, J. A., Ho, N. T., and Ho, C. (2005) A biophysical investigation of recombinant hemoglobins with aromatic B10 mutations in the distal heme pockets. *Biochemistry* 44, 7207–7217.
- (46) Brucker, E. A. (2000) Genetically crosslinked hemoglobin: A structural study. *Acta Crystallogr. D56*, 812–816.
- (47) Mathews, A. J., Rohlfs, R. J., Olson, J. S., Tame, J., Renaud, J. P., and Nagai, K. (1989) The effects of E7 and E11 mutations on the kinetics of ligand binding to R state human hemoglobin. *J. Biol. Chem.* 264, 16573–16583.
- (48) Esquerra, R. M., Lopez-Pena, I., Tipgunlakant, P., Birukou, I., Nguyen, R. L., Soman, J., Olson, J. S., Kliger, D. S., and Goldbeck, R. A. (2010) Kinetic spectroscopy of heme hydration and ligand binding in myoglobin and isolated hemoglobin chains: An optical window into heme pocket water dynamics. *Phys. Chem. Chem. Phys.* 12, 10270–10278.
- (49) Park, S. Y., Yokoyama, T., Shibayama, N., Shiro, Y., and Tame, J. R. (2006) 1.25 Å resolution crystal structures of human haemoglobin in the oxy, deoxy and carbonmonoxy forms. *J. Mol. Biol.* 360, 690–701.
- (50) Schotte, F., Soman, J., Olson, J. S., Wulff, M., and Anfinrud, P. A. (2004) Picosecond time-resolved X-ray crystallography: Probing protein function in real time. *J. Struct. Biol.* 147, 235–246.
- (51) Cassoly, R., and Gibson, Q. (1975) Conformation, cooperativity and ligand binding in human hemoglobin. *J. Mol. Biol.* 91, 301–313.
- (52) Olson, J. S., and Phillips, G. N. (1997) Myoglobin discriminates between O₂, NO, and CO by electrostatic interactions with the bound ligand. *J. Biol. Inorg. Chem.* 2, 544–552.